

Hydrogen Chloride Optical Gas Standards (OGS) at PTB

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Summary:

Accurate and reliable measurements of hydrogen chloride (HCl) are required in applications such as biomethane quality control, clean room monitoring or stack emissions monitoring. In order to perform HCl measurements, standardized measurement methods as well as accurate reference gases are required to calibrate typical HCl measurement instruments. However, there is a lack of SI-traceable HCl reference gases and reliable quality control test methods for many of these applications, once future more challenging limit values have been put into force by new regulations to come. To this end, PTB is developing optical gas standards (OGS), e.g., for HCl quantifications in those applications and to complementing existing gaseous reference standards. In this paper, we report on the HCl-OGSs developed in PTB for different applications.

Keywords: Metrology, Gas Analysis, Optical Gas Standard (OGS), TILSAM, dTDLAS.

1. Introduction

Gaseous hydrogen chloride (HCl) poses severe health effects when inhaled and can form corrosive hydrochloric acid on surfaces when it meets water. These properties accelerate the need for accurate HCl detection e.g. for quality control measurements in biomethane, airborne molecular contaminations monitoring in clean rooms or stack gas emissions [1]. Accurate HCl measurements typically require validated test methods [1]. Reliable test methods are lacking for HCl quantifications in biomethane, an energy gas that is seen to replace parts of the fossil natural gas sources in existing grids [1]. For stack emissions, HCl measurements are referred to the HCl - standard reference method described in EN 1911 (on the determination of mass concentration of gaseous chlorides). EN 1911 is based on wet chemistry. Hence, a gas sample is extracted, particle-filtered and dried and then dissolved in water, to analyze the Cl⁻ ion concentration in the liquid. This sampling procedure can easily lead to systematic deviations. HCl amount fractions in biomethane and clean room air are required to stay at low $\mu\text{mol/mol}$ to the nmol/mol levels. HCl sensor calibration requires calibration gas standards in the same range. However, generation and provision of gaseous reference materials traceable to the international system of units (SI) has proven to be difficult [1], e.g. there are no calibration and measurements capabilities (CMCs) reported for HCl amount fractions below $10 \mu\text{mol/mol}$ (<https://kcdb.bipm.org/>). Only a

few National Metrology Institutes (NMIs) have CMCs for HCl ($10\text{-}1000 \mu\text{mol/mol}$) in N_2 . For HCl in more complex gas matrices (e.g. biomethane), there are no CMCs available at all. Optical gas standards [3], [4] provide the option to be mandated as test methods for HCl measurements in the above mentioned applications. Due to the 1st principles measurement approach, OGSs do not require calibration with a calibration gas mixture, and therefore can be used to complement gaseous reference standards in the low $\mu\text{mol/mol}$ down to the nmol/mol levels [2, 3]. Furthermore, an OGS can also be used for HCl measurements directly in the field in situ. In this paper, we present HCl OGS instruments compliant with the TILASM method [4] and developed or currently being developed at PTB.

2. An Optical Gas Standard (OGS)

The measurement technique employed in an OGS instrument (see Fig. 1) is direct tunable diode laser absorption spectroscopy (dTDLAS) [2-3], [5-6]. dTDLAS is a variant of TDLAS that combines TDLAS with a 1st principles data evaluation approach to derive absolute gas species amount fractions that are directly traceable to the SI. An OGS laser spectrometer is thus similar to the National Institute of Science and Technology (NIST) ozone standard reference Photometer (SRP). Employing the Beer-Lambert law on a continuously scanned diode laser spectrometer and deriving the line area (A_{line}) underneath an absorption line, a SI-

traceable HCl amount fraction is inferred. Figure 1 shows an HCl absorption profile in N₂ matrix gas measured at a wavelength of about 3.6 μm and evaluated according to equation 1 as resulting a 518.4 μmol/mol HCl amount fraction without the need to calibrate the instrument with a gaseous calibration standard [3], [4]. Repeated measurements are shown in Fig. 2.

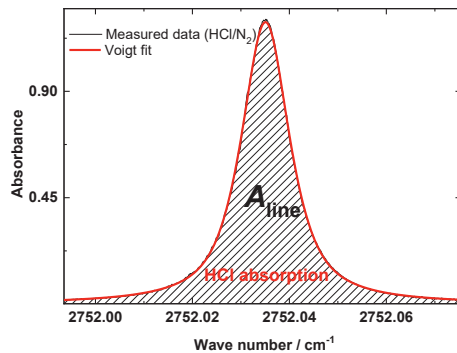


Fig. 1: Typical HCl single line absorbance spectrum measured by an OGS instrument operated at 3.6 μm.

$$x_{\text{species}} = \frac{k_B \cdot T}{S_T \cdot L \cdot p_{\text{total}}} \cdot A_{\text{line}} \quad (1)$$

For an HCl OGS, ensuring that all input quantities on the right-hand side of Eq. 1 are SI-traceable, the HCl amount fraction (concentration) x_{HCl} is directly traceable to the SI. The quantities k_B being the Boltzmann constant, S_T the line strength of the probed molecular transition at gas temperature T , L the path length of the light beam transmitted through the absorbing medium and p_{total} the total gas pressure. The SI-traceability of these input quantities is ensured similarly to references [2,3,4].

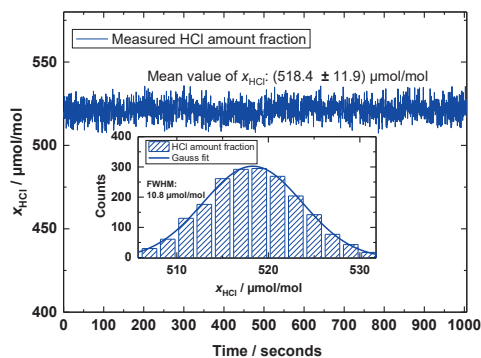


Fig. 2: HCl amount fraction (repeated measurements) as a function of time. Inset: A histogram depicting a normal distribution of the results and depicting the performance of an OGS.

Delivering SI-traceable amount fractions, an OGS is a “calibration free” instrument and can therefore be used as described to complement (support and use in the place of) calibration gases both in the lab and in the field, especially for sticky and reactive gases such as HCl, H₂O and NH₃.

3. Summary

Table 1 lists some details of HCl OGS systems developed/currently being developed at PTB. The OGS-Biomethane instrument is currently being used in a bilateral comparison with the Korean NMI, KRISS. Furthermore, the instrument will be employed in a CCQM key comparison of HCl in air, planned to be run at a 30 μmol/mol level.

Table 1: Summary of HCl OGS systems developed/currently being developed at PTB.

OGS systems of PTB cluster	Targeted range / μmol/mol	Rel. combined uncertainty ($k = 1$) / %	Gas Matrix
OGS-Biomethane	0.025 - 500	2.3	CH ₄ (or biomethane), N ₂ , air
OGS-Stack	0.05 - 100	2 - 4	Flue gas (with high CO ₂ , H ₂ O)
OGS-AMC	0.001 - 1	2 - 4	Air (or N ₂)

OGS-Biomethane: HCl-OGS instrument for bio-CH₄ conformity assessments.
OGS-Stack: HCl-OGS instrument for stack emissions monitoring.
OGS-AMC: HCl-OGS instrument for airborne molecular contaminations tests.

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References

- [1] EMPIR project Biomethane (<http://empir.npl.co.uk/biomethane/>), IMPRESS 2 (<http://empir.npl.co.uk/impress/>), MetAMCII, (<http://empir.npl.co.uk/metamcii/>).
- [2] J. A. Nwaboh, Z. Qu, O. Werhahn and V. Ebert, "Interband cascade laser-based optical transfer standard for atmospheric carbon monoxide measurements", *Applied Optics*, 56, E84-E93 (2017).
- [3] B. Buchholz, N. Böse and V. Ebert, "Absolute validation of a diode laser hygrometer via inter-comparison with the German national primary water vapor standard", *Applied Physics B*, 116, 883–899 (2014).
- [4] O. Werhahn, J.C. Petersen (eds.), 2010, TILSAM technical protocol V1_2010-09-29. Available from: http://www.euramet.org/fileadmin/docs/projects/934_METCHEM_Interim_Report.pdf.
- [5] Z. Qu, J. Nwaboh, O. Werhahn and V. Ebert, "Towards a dTDLAS-Based Spectrometer for Absolute HCl Measurements in Combustion Flue Gases and a Better Evaluation of Thermal Boundary Layer Effects", *Flow, Turbulence and Combustion*, Sept. 2020, doi: 10.1007/s10494-020-00216-z.
- [6] A. Pogany, S. Wagner, O. Werhahn and V. Ebert, "Development and Metrological Characterization of a Tunable Diode Laser Absorption Spectroscopy (TDLAS) Spectrometer for Simultaneous Absolute Measurement of Carbon Dioxide and Water Vapor", *Applied spectroscopy*, 69, 2, 257-268 (2015).