

limited. The only solvent that can be used at room temperature is hexafluoroisopropanol (HFIP).

In order to protect the copper wires and the sensor from direct contact with the coating material, the guide wire is coated with a thin polyester shrink tubing.

The first experimental results show that the polyester shrink tubing is being damaged by the coating material, so that the wires are not further protected and are in fact in contact with the pebax. Chemically the raw pebax is not aggressive to plastics, however there is no information hereof for the solvent HFIP, with which it is liquefied. It is especially interesting to see how it affects polyurethane. In case of incompatibility with polyurethane, it may damage the wire insulation (Figure 2) and consequently come to electrical short. Since HFIP is used to solve its aggressiveness to plastics, especially to polyurethane, needs to be investigated.

Through this work, the interaction between the coating material pebax and the wire insulation is to be investigated. Afterwards its applicability should be evaluated for the mentioned guide wire.

Implementation of Experiment

In this study the following experiments were performed:

Experiment 1: The aim of this experiment is to see, if HFIP solvent damages the insulation of the wires. Here, the copper wires were first immersed for different period of time (5 minutes, 5 hours and 3 days) in the pure solvent (HFIP) (Figure 3, left). Then they were immersed in a solvent of nickel sulfate μ Chem 410 (SurTec Germany GmbH). By applying a current of 2mA nickel can be deposited in the electroplating bath at the places, where the insulation was damaged (Figure 3, right).

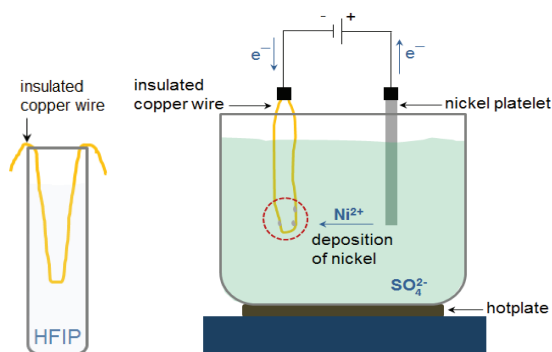


Fig. 3: Implementation of experiment, Left: insulated copper wire immersed in HFIP, Right: galvanizing of copper wire

Experiment 2: The aim of the second experiment is to investigate the influence of the dissolved Pebax on the wire insulation. The wires were wound around a stainless steel core and immersed three times in a pebax bath. Each immersion lasted 10 seconds and was followed by a one minute drying period. In order to test, if the insulation of the copper wires were damaged electrical short measurements were conducted, Figure 4.

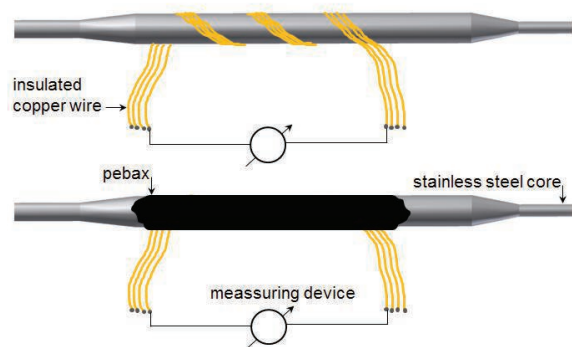


Fig. 4: Implementation of experiment, Above: copper wires coiled around a stainless steel core; Below: coated copper wires with pebax and check for short circuit

Experimental Results

The first experimental results show that the degree of damage of the wire insulation increases with the immersion time of the wires in the HFIP solvent. Figure 5 shows the microscope images of the galvanized copper wires. It can be clearly seen that the insulation of wire_1 (Figure 1a) was damaged more than wire_2 (Figure 1b). This can be recognized by the deposited nickel at the damaged places. In contrast the insulation of wire_2 (Figure 1c) was not damaged.

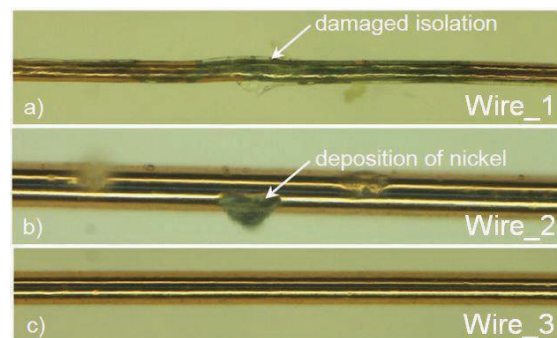


Fig. 5: Galvanized copper wires with different immersing time in HFIP, a) 3 days, b) 5 hours, c) 5 minutes

In order to evaluate the influence of pebax coating on the wiring, the current-voltage curve was recorded. In Figure 6 the current-voltage diagrams of the copper wires, shown in Figure 5 during the galvanizing are given. By the

current-voltage diagrams the damage of the wire insulation revealed in Experiment 1 could be confirmed.

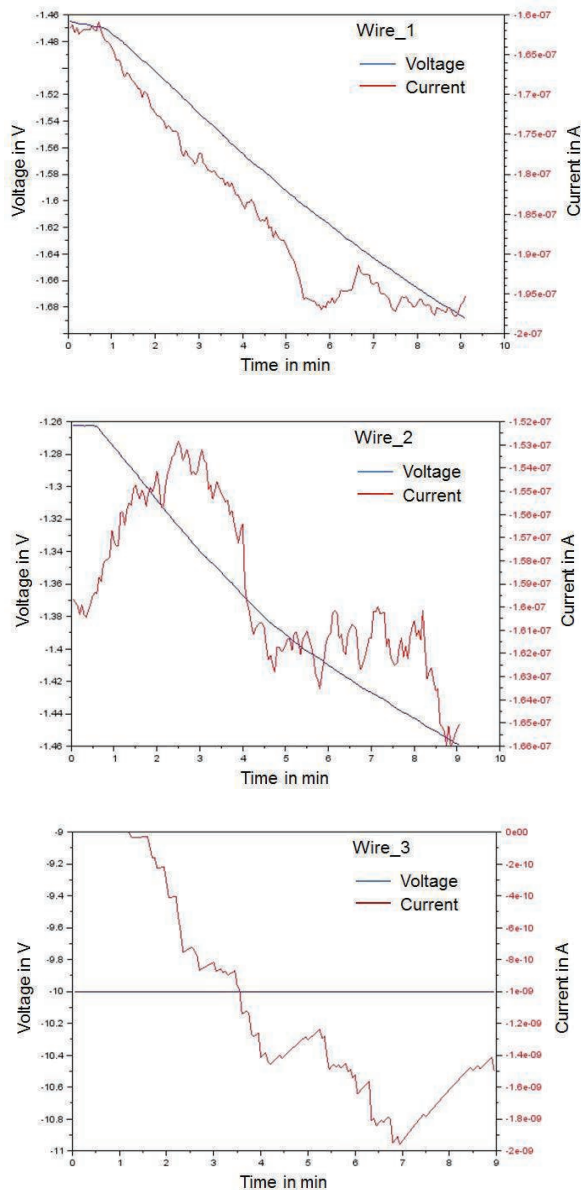


Fig. 6: Current-Voltage diagrams of copper wires during galvanization

As expected, the measured voltage in wire_1 was higher than in wire_2. The reason is that the wire insulation of both wires was damaged differently, due to the different duration of the immersion. The voltage increase over the time (see Figure 6, wire_1 and wire_2) can be explained by the deposited nickel layer on the damaged areas. Wire_3 voltage couldn't be measured because the wire insulation is intact.

The results of the second experiment show that the coating material is aggressive towards the polyester protective tube, but has no influence on the wire insulation. The reason is the high chemical resistance of the polyamide insulation,

and the short interaction time due to the rapid evaporation of HFIP. Also, the coating viscosity does not influence the degree of damage of the wire insulation.

Conclusions

From the current-voltage diagram it can be seen that depending of the immersing time pure HFIP solvent caused damage to the sensor wiring. From the short time of immersion in the dissolved pebax the insulation of the copper wires was not damaged. In this study it could be proved that pebax 3533 can be used as a coating material for medical guide wires.

The next step is to investigate the influence of pebax on the force sensor by comparing the sensor measurement signal before and after coating with pebax.

Acknowledgements

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References

- [1] Meiss, T.: *Konstruktion eines Mikrokraftsensors für Herzkatheterisierungen*. Institut für Elektromechanische Konstruktion, Dissertation, TU Darmstadt, 2012.
- [2] Meiss, T.; Werthschützky, R.: *Konstruktion eines Mikrokraftsensors für Herzkatheterisierungen*. Technisches Messen, Oldenburg Verlag, Vol. 76, Iss. 6, pp. 292-299.
- [3] Elsner, P.; Eyerer, P.; Hirth, Th.: *Kunststoffe: Eigenschaften und Anwendungen (Domininghaus)*. 8. neu bearbeitete und erweiterte Auflage, Springer Verlag, 2012, ISBN-10: 3642161723.