

# Factors affecting the electrical conductivity of TiO<sub>2</sub>-based gas sensors

Azhar Ali Haidry<sup>1,2,\*</sup>, Linchao Sun<sup>1,2</sup>, Zhong Li<sup>1,2</sup>, Lijuan Xie<sup>1,2</sup>, Qawareer Fatima<sup>1,2</sup>, Zhengjun Yao<sup>1,2</sup>

<sup>1</sup>College of Materials Science and Technology, Nanjing University of Aeronautics and Astronautics, 211100 Nanjing, China

<sup>2</sup>Key Laboratory of Materials Preparation and Protection for Harsh Environment, Ministry of Industry and Information Technology, 211100 Nanjing, China

\*Corresponding author's e-mail address: [aa.haidry@nuaa.edu.cn](mailto:aa.haidry@nuaa.edu.cn), [azharaliq2@gmail.com](mailto:azharaliq2@gmail.com)

## Abstract:

Polycrystalline TiO<sub>2</sub> submicron-grains with only rutile phase were synthesized via cost-effective thermal oxidation method under high temperature annealing. The gas sensors on these submicron-grains show excellent gas sensing properties in terms of their sensitivity, stability and reaction times. In addition to excellent sensing characteristics the sensors also show strong dependence of n/p type conductivity inversion various measurement conditions such as gas concentration, operating temperature and applied voltage. Further, the effect of surface modification by Ag, Ni, Pt and Au thin film on structural, morphological and gas sensor characteristics is studied in detail. The current study offer novel sensing phenomena that can be exploited to tailor the selectivity of the sensors.

**Key words:** TiO<sub>2</sub>, thermal oxidation, conductivity inversion, surface modification, operating conditions.

## Introduction

Metal Oxide (MOX) Semiconductors have attracted the attention of many scientists and engineers in the past few decades to be used as chemical sensors owing to their unique characteristics and simple operation, cost-effective fabrication and wide applicability. The demand for these chemical sensors has been growing continuously due to the emergence of new possible applications in a variety of industries. Due to the availability of modern high-tech equipment and the recent developments, the field of gas sensors can particularly benefit in the field of monitoring processes at high and harsh environments [1].

Among other MOX materials, titanium dioxide (TiO<sub>2</sub>) is a potential candidate for gas sensing applications due its excellent chemical stability, nontoxicity and innocuousness nature, low-cost and rich earth abundance. Despite these advantages, TiO<sub>2</sub>, however, suffer from poor selectivity and high operating temperature issues [2]. In addition, it is demonstrated that TiO<sub>2</sub> when operated under elevated conditions shows anomalous electrical conductivity behavior. In addition, TiO<sub>2</sub> exhibit unintentional conductivity (either n-type or p-type) in their as-grown state and lack of control over this key property has been a great impediment to

electronic applications. For instance, the common unintentional n-type conductivity in TiO<sub>2</sub> films has been widely attributed to oxygen vacancies. A solid basis for this hypothesis has been lacking, however. In the other case, it has now been established that the oxygen vacancies that were most often blamed for the observed n-type conductivity do not act as shallow donors. Instead, first-principles electronic structure calculations have suggested that the conductivity is due to unintentional incorporation of impurities, with hydrogen being a very likely candidate [3,4].

Depending on the circumstances, many factors can be considered to influence the charge carrier density ( $[e^-] \sim 10^{16} - 10^{18} \text{ cm}^{-3}$ ) and the mobility value ( $\mu \sim 10^2 - 10^3 \text{ cm}^2/\text{Vs}$ ), including grain boundaries, ionized impurities and electron-phonon interactions [5]. In the current paper, we found that the factors, such gas concentration, operating temperature and applied voltage, can significantly alter the sensor signal if not controlled well.

## Experimental

To begin with, high purity Ti-discs (99.9% purity) having diameter of  $\phi = 15 \text{ mm}$  and thickness of 1 mm, were washed and polished using standard metallurgical process to remove oil. Then the discs were coated with 30 nm films

of Ti, Ni, and Ag using RF magnetron sputtering technique. Finally, nanostructured TiO<sub>2</sub> films were obtained via thermal oxidation under static air condition in a tube furnace OTF-1200X tubular furnace at 1000 °C for 1 hour. Subsequently, the interdigitated electrodes with a spacing of 10 μm were fabricated by photolithography method. The schematic diagram of overall preparation steps is shown in Fig. 1. The sensors were further characterized with XRD, SEM, TEM and gas sensors tests as mentioned in previous reports elsewhere [6,7].

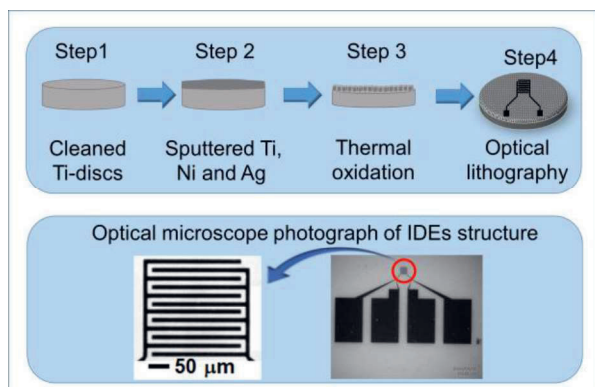


Fig. 1. The schematics of sensor fabrication process.

## Results

The sensors prepared in this work show excellent sensitivity towards H<sub>2</sub> with its selective monitoring thus making it a potential candidate for industrial applications such as fuel-cell. From the calibration of the curves obtained in Fig. 2 which shows the sensor response relationship with increased H<sub>2</sub> concentration at operating temperature 200 °C, it is found that the change in electrical resistance  $\Delta R$  (sensor response) is expressed by a power law relation between gas concentration and sensor response ( $\Delta R \sim kC^a$ ).

In addition, we found strong co-relation of gas concentration, operating temperature and applied not only with the sensor response but also with electrical conductivity of pure and modified TiO<sub>2</sub>. Based on these facts, we will further explore the sensing and electronic transport mechanism to fully understand aforementioned factors.

## References

- [1] N. Barsan, D. Koziej, U. Weimer, Metal oxide-based gas sensor research: How to, *Sensors and Actuators B* 121 (2007), pp. 18 – 35. <https://doi.org/10.1016/j.snb.2006.09.047>
- [2] A. Gurlo and R. Riedel, In Situ and operando spectroscopy for assessing mechanisms of gas

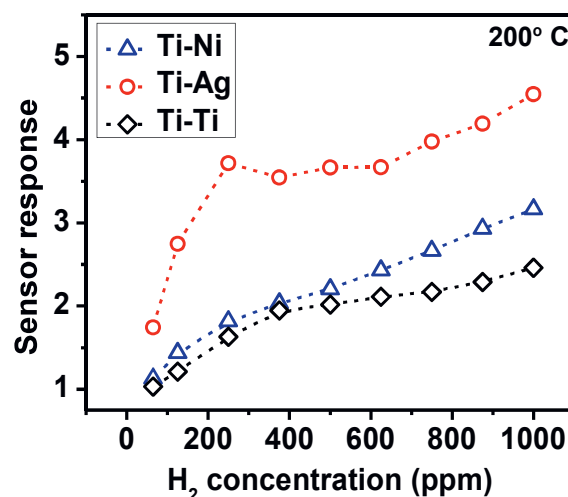


Fig. 2. The sensor response relationship with increased H<sub>2</sub> concentration at operating temperature 200 °C.

sensing, *Angew. Chem. Int. Ed* 2007, 46, pp. 3826 – 3848. DOI: 10.1002/anie.200602597

- [3] A. A. Haidry, Cagdas Cetin, Klemens Kelm, Bilge Saruhan, Sensing mechanism of low temperature NO<sub>2</sub> sensing with top–bottom electrode (TBE) geometry, *Sens. Actuators, B*, 2016, 236, 874-884. <https://doi.org/10.1016/j.snb.2016.03.016>
- [4] N. Yamazoe, K. Shimano, Theory of power law for metal oxide, *Sens. Actuators, B*, 2008, 128, 566- 573. <https://doi.org/10.1016/j.snb.2007.07.036>
- [5] A. A. Haidry, A. A. Haidry, A. Ebach-Stahl, B. Saruhan, Effect of Pt/TiO<sub>2</sub> interface on room temperature hydrogen sensing performance of memristor type Pt/TiO<sub>2</sub>/Pt structure, *Sensors and Actuators B: Chemical* 253, 1043-1054. <https://doi.org/10.1016/j.snb.2017.06.159>
- [6] L. Sun, A. A. Haidry, Q. Fatima, Z. L and Z. Yao, Improving the humidity sensing below 30% RH of TiO<sub>2</sub> with GO modification, *Mater. Res. Bull.*, 2018, 99, 124-131. <https://doi.org/10.1016/j.materresbull.2017.11.001>