

Label-free detection of Potassium ion using PEDOT:PSS

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Abstract

Potassium ion is an important biomarker that plays an important role in heart. Normal potassium levels are between 3.5 and 5.0 mmol/L. Currently, there are methods of attaching guanine-rich DNA to the electrode surface or using potentiometry to detect potassium. However, this method of using DNA has disadvantages of being unstable due to DNA. A disadvantage of potentiometry is that a chamber must be made for the reference electrode. Therefore, we propose a detection mechanism for measuring potassium using the PEDOT:PSS property [1]. RGO-PEDOT:PSS was spin-coated at 1200 rpm for 60 s on ITO glass. ITO spin-coated, Ag/AgCl electrode, Pt electrode were used. The reaction was confirmed by cyclic voltammetry according to the concentration of KCl (0.1 mM–10 mM KCl). As shown in Fig. 1, in the C-V curve of D.D.I water does not occur a peak that indicates a redox reaction. On the other hand, the C-V curves of KCl occur a peak voltage about 0.35 V. Their currents increase with higher concentration as shown in Fig. 1.

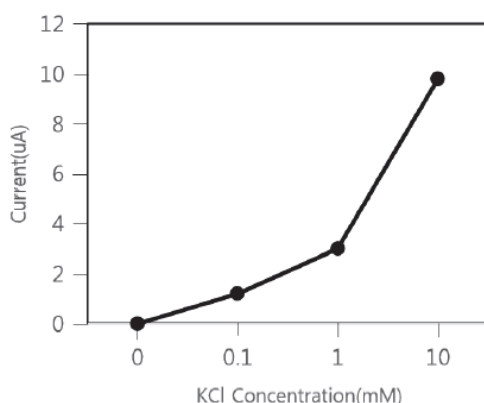


Fig. 1. Current of KCl using RGO-PEDOT:PSS

Key words: KCl, RGO-PEDOT:PSS, Electrochemical sensor, Cyclic voltammetry

References

[1] Park, HS, Ko SJ, Park JS, Kim JY, Song HK, *Scientific reports* 3, 2454-2459(2013)
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