# Zinc Oxide Dosimeter-type NO<sub>2</sub> Sensor Prepared by Discontinuous Powder Aerosol Deposition

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# **Summary:**

 $NO_2$  dosimeters operated at room temperature provide an opportunity to detect the  $NO_2$  concentration in ambient air and, specifically, directly mean values  $NO_2$ . It is evaluated whether the functional material, Al-doped zinc oxide, can also be applied by a room temperature process, the discontinuous powder aerosol deposition. The sensors are subsequently characterized with respect to their  $NO_2$  dosimeter behavior at room temperature. Additionally, the influence of a low temperature annealing step of the DPAD-films on the  $NO_2$ -dosimeter characteristics is investigated.

**Keywords:** dosimeter-type sensor,  $NO_2$  detection, zinc oxide, powder aerosol deposition, room temperature device

# Dosimeter-type NO<sub>2</sub>-sensor

NO<sub>2</sub> detection in ambient air is still challenging, especially concentrations in the range of ppb. For air quality control the detection of the hourly mean values or the total amount (dose) of NO2 over a certain time interval are important. The working principle of a dosimeter-type NO2 sensor operating at room temperature is described in [1] and [2]. The gas dosimeter detects the total dose of NO2, i.e., the timely integration of the concentration over a certain time span, directly. The impedimetric device is based on sol-gel-synthesized Alumina-doped Zinc oxide (Al:ZnO) as functional material applied in thick-film technology [3]. During the first operation phase, the sorption phase, the NO<sub>2</sub> molecules are sorbed and the measured resistance changes almost linearly with the sorbed NO<sub>2</sub> amount. In absence of NO<sub>2</sub>, the resistance remains constant. After reaching a certain loading state, which means that the adsorption sites are occupied, an UV-initiated regeneration phase follows. It releases all sorbed NO<sub>2</sub>-species. Afterwards, a new loading cycle starts. The dosimeter allows for the detection of NO<sub>2</sub> at ppb-levels and therefore also the resulting hourly mean values.

#### **Discontinuous Powder Aerosol Deposition**

In contrast to the preparation of screen-printed thick films with consequent sintering steps at higher temperatures, the (discontinuous) powder aerosol deposition ((D)PAD) allows to prepare ceramic films at room temperature [4]. The ceramic powders are aerosolized by a carrier gas flow and accelerated into a vacuum chamber by a pressure gradient (Fig. 1). The impact of the

particles leads, depending on the kinetic energy of the particles, to the formation of dense ceramic films on the substrates. Powder quantities below 100 mg can be deposited by DPAD. The nanocrystalline structure with many grain boundaries and the distorted lattice of the as deposited film lead to a high resistance of the film. Mild annealing (far below the sintering temperature) allows for restoring the reduced electric conductivity close to bulk values again [4,5].

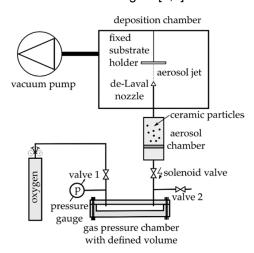


Fig. 1. Scheme of discontinuous powder aerosol deposition DPAD [4]

For a room temperature sensor device like a Al:ZnO-NO<sub>2</sub> dosimeter, a preparation technology working at room temperature would be advantageous. Therefore, sol-gel synthesized Al:ZnO-powders were deposited as sensing layers by DPAD and the dosimeter-type sensing characteristics of the films were investigated.

## **Experimental**

Al:ZnO powder with 5% Al was synthesized as descried in [3]. The powder is then deposited by DPAD on top of platinum-interdigitated electrodes (IDE). To cover the IDE-area, three circular shaped films of Al:ZnO are applied. The resulting sensor device consists of a platinum heating structure on an alumina substrate and the Pt-IDE covered with the Al:ZnO-film.

The sensor behavior was investigated in dry synthetic air with NO<sub>2</sub> concentrations between 0 and 200 ppb at a constant flow rate. The impedance of the sensors was measured ( $U_{\rm eff}$  = 100 mV, f = 1 Hz) and the resistance was calculated assuming an R||C equivalent circuit. Before and after each measurement, the sensor was regenerated by UV-light exposure (385 nm). Additionally, the influence of thermal annealing of the Al:ZnO-films from 100 to 400 °C on the dosimeter signal (measured at RT) was characterized.

## **Results and Discussion**

The relative resistance change  $(R-R_0)/R_0$  ( $R_0$ : resistance in base gas; R: resistance with NO<sub>2</sub>), measured at room temperature, during exposure to four NO<sub>2</sub> pulses are shown in Fig. 3. Indicated are the thermal annealing temperatures that are applied to the films. The sensor signal increases linearly for all annealing temperatures during NO<sub>2</sub> exposure and remains constant during NO<sub>2</sub>-pauses. The sensor shows typical dosimeter-type behavior. An influence of the annealing temperature on the dosimeter behavior can be observed.

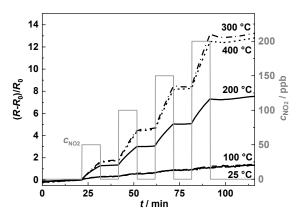


Fig. 3. Dependence of the relative resistance change  $(R-R_0)/R_0$  on the  $NO_2$ -amount and the annealing temperature measured at room temperature

Fig: 4 shows the characteristic dosimeter curves,  $(R-R_0)/R_0$  versus the calculated NO<sub>2</sub>-dosis (in ppb s). Compared to screen-printed films, as expected, the DPAD-films show a lower NO<sub>2</sub>-sensitivity. A linear relationship is visible, and the sensitivity increases with increasing annealing temperature. It is expected that the observed sensitivity increase is due to occurring oxygen

desorption processes and additionally effects due to the thermal annealing of the film (relaxation of stresses, morphology changes). These processes will be investigated more in detail.

#### Conclusion

The work shows, that DPAD-deposited Al:ZnO films are sensitive to  $NO_2$  at room temperature and provide dosimeter-type behavior. An influence of a thermal annealing on the dosimeter properties is observed which will be investigated in the future. Additionally, the effect of humidity needs to be addressed.

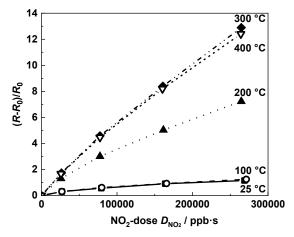


Fig. 4. Relative resistance change  $(R-R_0)/R_0$  versus calculated dose of  $NO_x$  in dependence of sensor annealing temperature

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