Thin-Film Gas Sensors Operating in a Perpendicular Current Mode

Yurie Ishikawa, Kazuhiro Hara Tokyo Denki University, 5 Senju-Asahi-cho, Adachi-ku, Tokyo 120-8551 Japan hara@eee.dendai.ac.jp

Abstract

Thin-film gas sensors operating in a perpendicular current mode are presented. To achieve this mode, a mesh-like top electrode is deposited on a sensing film instead of an electrode that is normally deposited on a substrate. The sensor current flows perpendicularly to the surface of the thin film and thus is not blocked by the cracks that are generated by an annealing process after deposition and by the difference of the thermal expansion coefficients between the sensing film and the substrate. The sensor is expected to work for a longer period of time without possible deterioration caused by cracks. Moreover, the mesh-like top electrode enhances the sensitivity to hydrogen by acting as a catalyst when Pt is used for the top electrode.

Key words: hydrogen sensor, thin film, metal oxide, cracks, perpendicular current mode

Introduction

Thin-film gas sensors tend to deteriorate with an increase in cracks for SnO_2 -based [1][2] and WO_3 -based films[3]. These cracks are usually generated perpendicular to the surface of the sensing film by the difference of the thermal expansion coefficients between the sensing film and the substrate.

Some cracks are also generated by an annealing process after deposition that is usually carried out for the recovery from various defects and the stabilization of the crystal structure. In this process metal-oxide films generally shrink as a consequence of sintering while its substrate usually expands as a consequence of thermal expansion. During an annealing process, the internal stress in the film increases and then cracks are generated to release the internal stress.

The current flow in the sensing film is schematically shown in Fig.1 for a traditional thin-film gas sensor operating in a parallel current mode. The sensor current is apparently blocked by the cracks. The increase in the number of cracks decreases the sensor current and thus its electrical conductance.

In this paper some examples of cracks observed by SEM are shown for a variety of sensor materials. Then a novel thin-film gas sensor that reduces the adverse effect of cracks is proposed. In this sensor the top electrode is deposited on the sensing film in

order to change the current direction as shown in Fig.2. The current flows perpendicularly to the surface from the bottom electrode to the top electrode without the adverse effects caused by the cracks.

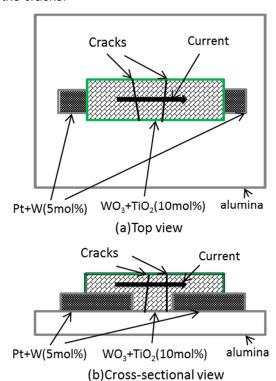
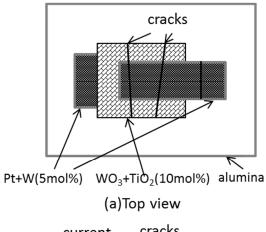


Fig.1 Traditional thin-film gas sensor operating in a parallel current mode.



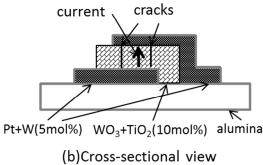


Fig.2 Proposed thin-film gas sensor operating in a perpendicular current mode.

Sensor fabrication

All films in Fig. 2 were successively deposited on an alumina substrate by r.f. sputtering. The thickness of the bottom Pt-based film, the WO₃based film and the top Pt-based film was 200nm, 100nm and 10nm, respectively. The sensor geometry defined was bv photolithography. Sensors with based($SnO_2+Pd(4mol\%)+V_2O_5(4mol\%)$), Fe_2O_3 -based(Fe_2O_3 +TiO₂(5mol%)+MgO(4mol%)), NiO and TiO₂ thin-films were also fabricated.

Cracks in thin-films for various materials

Figure 3 shows the surface morphology of a SnO_2 -based film. This film was annealed at $600^{\circ}\mathrm{C}$ in air soon after deposition. Two cracks are clearly observed.

Figure 4 shows the surface morphology of a NiO-based film. This film was annealed at 1000 $^{\circ}$ C in air after deposition. A crack is observed in addition to the crystal growth of the grains.

Figure 5 shows the surface morphology of a WO₃-based film. This film was heated at 420°C for 4 months after annealing at 600°C in air. Several cracks are also observed in addition to a structural irregularity of the morphology.

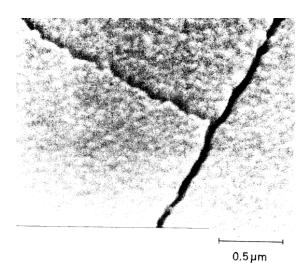


Fig. 3 Surface morphology of a SnO₂-based film after being annealed at 600 $^{\circ}$ C.

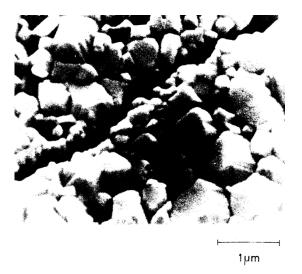


Fig. 4 Surface morphology of a NiO-based film after being annealed at 1000° C.

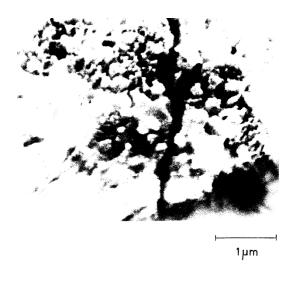


Fig. 5 Surface morphology of a WO₃-based film after operating at $420^{\circ}C$ for 4 months.

 WO_3 -based film peeled off from the substrate when it was annealed at a temperature higher than 600° C.

Figure 6 shows the surface morphology of a TiO_2 -based film. This film was heated at $420^{\circ}C$ for 4 months after annealing at $600^{\circ}C$ in air. Several cracks are also observed.

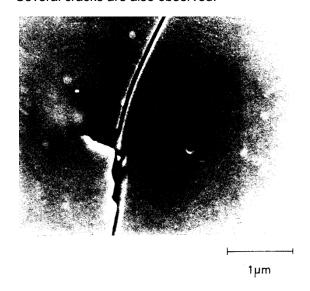


Fig. 6 Surface morphology of a TiO_2 -based film after operating at $420^{\circ}C$ for 4 months.

Sensing characteristics and discussion

WO₃-based, SnO₂-based and Fe₂O₃-based thin-film sensors were tested at 3000ppm of hydrogen gas. Figure 7 shows the temperature dependence of the sensitivity to hydrogen for these sensors. Here the sensitivity is defined as the ratio of the resistance in air to that in gas. The WO₃-based sensor shows considerably high sensitivity to hydrogen in the lower temperature range near 100 $^{\circ}\mathrm{C}$ though SnO₂-based and Fe₂O₃-based thin-film sensors are also sensitive to hydrogen gas.

The dependence of the sensitivity on hydrogen concentration is shown in Fig. 8 for a WO₃-based sensor. The sensor was capable of detecting hydrogen at a concentration as low as 100ppm.

The sensitivity to various gases is shown in Fig.9. The WO₃-based sensor was sensitive to hydrogen gas among various gases. Thus the selectivity to hydrogen gas was quite good.

As for the sensors with SnO₂-based film, long-term stability was examined. A traditional sensor operating in a parallel current mode degraded gradually and its conductance gradually decreased. In contrast, our novel sensor operating in a perpendicular current mode underwent less change.

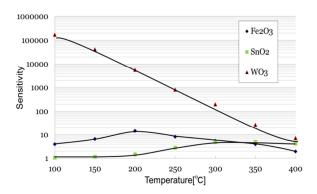


Fig.7 Temperature dependence of the sensitivity to hydrogen for WO_3 -based, SnO_2 -based and Fe_2O_3 -based thin-film sensors. Hydrogen concentration is 3000ppm.

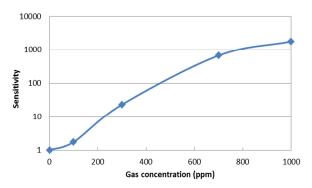


Fig.8 Dependence of the sensitivity on hydrogen concentration for WO_3 -based sensor at an operating temperature of $150^{\circ}C$.

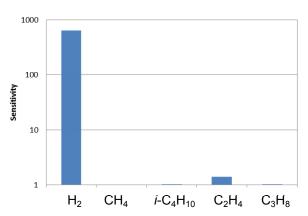


Fig.9 Sensitivity to various gases for WO₃-based sensor at an operating temperature of $150\,^{\circ}\mathrm{C}$.

The transient response for WO₃-based sensors is shown in Fig. 10 and 11, at an operating temperature of 100° C and 150° C, respectively.

The response time and recovery time that are taken at the time for the resistance to reach 90% of the change are 1sec and 22min, respectively at $100\,^{\circ}\mathrm{C}$. The recovery time is quite long while the response time is short.

The response time and recovery time are 22sec and 2min, respectively at 150° C. The recovery time is much short compared that at 100° C.

The thickness of the top electrode Pt was adjusted so that the film was mesh-like. Hydrogen gases adsorb on the surface of the sensing film that is not covered by Pt while they are activated by the catalytic action of Pt that exists nearby. Thus the sensitivity and selectivity to hydrogen are enhanced.

The temperature dependence of the sensitivity to hydrogen for 2 kinds of Fe₂O₃-based sensors with differnt thicknesses of the top electrode is shown in Fig. 12.

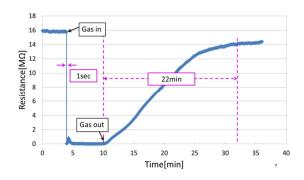


Fig.10 The transient response for WO_3 -based sensor at an operating temperature of $100^{\circ}C$.

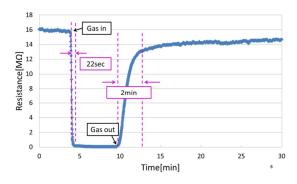


Fig.11 The transient response for WO_3 -based sensor at an operating temperature of 150 $^{\circ}C$.

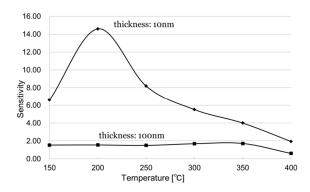


Fig. 12. Temperature dependence of the sensitivity to hydrogen for Fe_2O_3 -based sensors with different thicknesses of the top electrode: 10nm and 100nm.

When the thickness of the top electrode was 100nm, the sensitivity to hydrogen was low. In this sensor the surface of the sensing film was fully covered by the top electrode and thus there were no sites for hydrogen to adsorb. This is the reason for low sensitivity to hydrogen.

Conclusion

Thin-film gas sensors operating in a perpendicular current mode are presented. The sensor current flows perpendicularly to the surface of the thin film and thus is not blocked by the cracks. The sensors are expected to work for a longer period of time without possible deterioration caused by the cracks. Moreover, when Pt is used for the top electrode, by acting as a catalyst the sensitivity to hydrogen is enhanced.

References

- [1] J. Yu, Z.A. Tang, G.Z. Yan, P.C.H. Chan, Z.X. Huang, Sensors and Actuators B 139, 346-352 (2009); doi: 10.1016/j.snb.2009.03.033
- [2] R.K. Sharma, P.C.H. Chan, Z. Tang, G. Yan, I.-M. Hsing, Sensors and Actuators B 81, 9-16 (2001).
- [3] T. S. Kim, Y. B. Kim, K. S. Yoo, G. S. Sung, H.J. Jung, Sensors and Actuators B 62, 102-108 (2000).