

# H<sub>2</sub> Gas Sensor Based on PdO<sub>x</sub> doped In<sub>2</sub>O<sub>3</sub> Synthesized by Flame Spray Pyrolysis

Kanitha Inyawilert<sup>1</sup>, Anurat Wisitsoraat<sup>2</sup>, C. Liewhiran<sup>1</sup>, Sukon Phanichphant<sup>3,\*</sup>

<sup>1</sup> Department of Physics and Materials Science, Faculty of Science, Chiang Mai University, Chiang Mai 50202, Thailand

<sup>2</sup> Nanoelectronics and MEMS Laboratory, National Electronics and Computer Technology Center, National Science and Technology Development Agency, Klong Luang, Pathumthani 12120, Thailand

<sup>3</sup> Materials Science Research Center, Faculty of Science, Chiang Mai University, Chiang Mai 50202, Thailand

sphanichphant@yahoo.com\*

## Abstract:

0–1.0wt% Pd doped In<sub>2</sub>O<sub>3</sub> nanoparticles have been successfully synthesized in a single step by flame spray pyrolysis (FSP) technique using Indium (III) nitrate hydrate and palladium (II) acetylacetonate, as precursors dissolved in ethanol and their hydrogen sensing characteristics have been investigated. The particle and sensing film properties were analyzed by XRD, BET, TEM and XPS. The sensing films were prepared by spin coating technique. The crystallite sizes of In<sub>2</sub>O<sub>3</sub> spherical and cubic morphologies were found to be ranging from 2 to 20 nm and Pd might form solid solution with In<sub>2</sub>O<sub>3</sub> lattice. Gas-sensing characterization were studied at the operating temperatures ranging from 150 to 350°C in dry air. Hydrogen sensing characteristics of In<sub>2</sub>O<sub>3</sub> nanoparticles was significantly improved as Pd content increased from 0 to 1.0wt%. The 0.50wt%Pd doped In<sub>2</sub>O<sub>3</sub> sensing film showed an optimum H<sub>2</sub> response of 3,526 towards 1.0vol% H<sub>2</sub> concentration at 250°C operating temperature. In addition, PdO<sub>x</sub> doped In<sub>2</sub>O<sub>3</sub> sensing films exhibited good selectivity towards hydrogen with improving of response and response time.

**Key words:** Flame spray pyrolysis, PdO<sub>x</sub> doped In<sub>2</sub>O<sub>3</sub>, Gas sensor, H<sub>2</sub> sensor

## Introduction

Hydrogen gas is expected to become a green and renewable energy source in the future for several applications, including fuel cell vehicles, space crafts, automobiles, power generators and aircrafts [1]. This light and odorless gas is highly flammable; hydrogen gas leaks can result in disastrous consequences, such as explosion [2]. However, H<sub>2</sub> gas is highly flammable and explosive at volume concentration higher than 4% mixed with air, thus leading to growing demands for high performance H<sub>2</sub> sensors to detect its leakage [3]. In this study the Pd doped In<sub>2</sub>O<sub>3</sub> nanoparticles were synthesized by flame spray pyrolysis. The effect of Pd on gas sensing performances were systematically studied and optimized for selective detection of H<sub>2</sub> gas. At the same time, the range of interest for its detection is 150–10,000 ppm, allowing for early leakage warning and explosive indication. So it's essential to detect the content of H<sub>2</sub> to avoid more serious accidents at incipient faults.

## Materials and Methods

Firstly, 0–1.0wt%Pd doped In<sub>2</sub>O<sub>3</sub> nanoparticles were synthesized by flame spray pyrolysis, which was previously established by our group. For sensor fabrication, flame-made 0–1.0wt%Pd doped In<sub>2</sub>O<sub>3</sub> nanopowders were thoroughly mixed and ground with the binder solution. The resulting paste was spin coated on Al<sub>2</sub>O<sub>3</sub> substrates equipped with Au interdigitated electrodes to form a sensing film. The resulting substrates were annealed at 450°C for 2 h in an oven for binder removal prior to sensing test. The gas-sensing performances of all sensors were characterized towards H<sub>2</sub>, C<sub>2</sub>H<sub>2</sub>, C<sub>2</sub>H<sub>4</sub>, C<sub>2</sub>H<sub>4</sub>O, C<sub>2</sub>H<sub>5</sub>OH, H<sub>2</sub>S and NO<sub>2</sub> under atmospheric conditions by the standard flow through technique in stainless steel chamber at operate temperature in the range of 150–350°C.

## Results and discussion

Characterizations significantly confirmed Pd<sup>2+</sup> and Pd<sup>4+</sup> was formed solid solution with In<sub>2</sub>O<sub>3</sub>

lattice. Fig.1 shows TEM images with corresponding SAED pattern of pure  $\text{In}_2\text{O}_3$  and optimal Pd doped  $\text{In}_2\text{O}_3$  and highest Pd doping. It is seen that the spherical and cubic nanoparticles with the particle size were in the range of 2–20 nm. The corresponding SAED patterns display diffraction rings of polycrystalline  $\text{In}_2\text{O}_3$  nanoparticles having (211), (222), (400), (440) and (622) dominant planes in good agreement with the preferred crystallographic orientations previously seen from the XRD patterns while there was no plane associated with PdO or PdO<sub>2</sub> phases for Pd-doped  $\text{In}_2\text{O}_3$  nanoparticles. The gas sensing characteristics were showed in Figures 2–5. The effect of operating temperature ranging from 150 to 350 °C towards  $\text{H}_2$  response of  $\text{In}_2\text{O}_3$  nanoparticles with different Pd doping indicated that the 0.5 wt% PdO<sub>x</sub> doped  $\text{In}_2\text{O}_3$  sensor has the highest response at the optimum working temperature of 250 °C exhibited high response of ~3,526 to 1.0 vol%  $\text{H}_2$ . The response and recovery time were 2 s and 12 min respectively. The PdO<sub>x</sub> doped  $\text{In}_2\text{O}_3$  can improve sensor response, and decrease response and recovery time of sensor and exhibited good selectivity towards hydrogen gas.

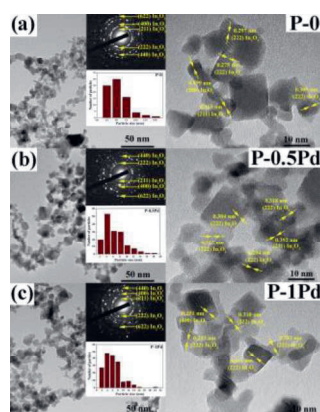


Fig.1. TEM images of the FSP-made (a) undoped  $\text{In}_2\text{O}_3$ , (b) 0.50 wt% Pd-doped  $\text{In}_2\text{O}_3$  and (c) 1.0 wt% Pd-doped  $\text{In}_2\text{O}_3$  nanoparticles. Insets: the corresponding SAED patterns and particle size distributions

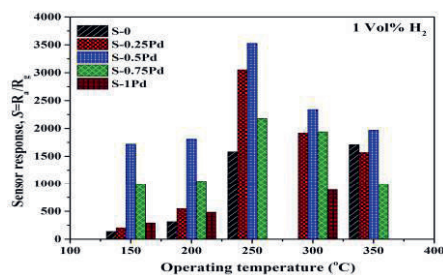


Fig.2. Histograms of sensor response to 1.0 vol% of  $\text{H}_2$  of FSP-made Pd-doped  $\text{In}_2\text{O}_3$  sensors with different Pd contents (S-0 to S-1Pd) at various operating temperatures ranging from 150 to 350 °C.

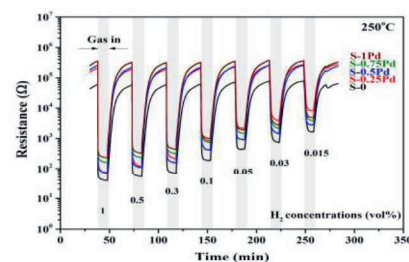


Fig.3. The change in resistance of  $\text{In}_2\text{O}_3$  sensing films with different Pd doping concentrations under exposure to various  $\text{H}_2$  concentrations ranging from 0.015 to 1.0 vol% at the operating temperature of 250 °C.

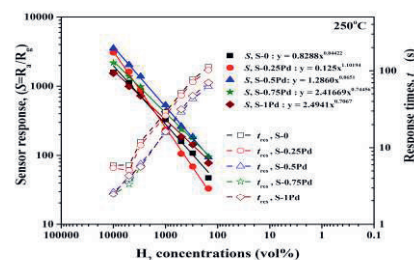


Fig.4. The corresponding sensor response ( $S$ ) (left axis) and response time ( $t_{res}$ ) (right axis) vs.  $\text{H}_2$  concentration of 0–1.0 wt% Pd doped  $\text{In}_2\text{O}_3$  sensors at 250 °C.

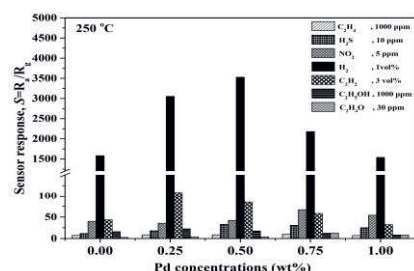


Fig.5. Selectivity histograms: sensor response of 0–1.0 wt% Pd doped  $\text{In}_2\text{O}_3$  sensors to 1,000 ppm  $\text{C}_2\text{H}_4$ , 1,000 ppm  $\text{C}_2\text{H}_5\text{OH}$ , 3.0 vol%  $\text{C}_2\text{H}_2$ , 1.0 vol%  $\text{H}_2$ , 10 ppm  $\text{H}_2\text{S}$  and 5 ppm  $\text{NO}_2$  at the optimal operating temperatures of 250 °C.

## References

- [1] K. Hacatoglu, I. Dincer, M.A. Rosen, Sustainability assessment of a hybrid energy system with hydrogen-based storage, *Int. J. Hydrogen Energy*, 40 (2015) 1559–1568; doi: 10.1016/j.ijhydene.2014.11.079.
- [2] R. Chen, X. Ruan, W. Liu, C. Stefanini, A reliable and fast hydrogen gas leakage detector based on irreversible cracking of decorated palladium nanolayer upon aligned polymer fibers, *Int. J. Hydrogen Energy* 40 (2015) 746–751; doi: 10.1016/j.ijhydene.2014.11.026.
- [3] L. B. Brett, J. Bousek, G. Black, P. Moretto, P. Castello, T. Hubert, U. Banach, Identifying performance gaps in hydrogen safety sensor technology for automotive and stationary applications, *Int. J. Hydrogen Energy*, 35 (2010) 373–384; doi: 10.1016/j.ijhydene.2009.10.064.