## Enhanced NO<sub>2</sub> response of surface decorated WO<sub>3</sub> lowcost gas sensors

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## Abstract:

The aim of this work is to enhance the electrical responses of low-cost WO $_3$  chemo-resistive gas sensors to sub-ppm NO $_2$  gas concentration through doping with different concentrations of Y $_2$ O $_3$ . Y $_2$ O $_3$  surface decorated WO $_3$  gas sensors were prepared by screen-printing and annealed at 800 °C. X-ray Diffractometry (XRD) and Scanning Electron Microscopy (SEM) were used to study the phase evolution and morphology of fabricated films, respectively. Results showed that Y $_2$ O $_3$  nanoparticles are well dispersed within the bulk WO $_3$  and have decorated the surface of WO $_3$  grains. Electrical measurements at 350°C operating temperature, showed that the relative responses (RR) of fabricated sensor to 200 ppb NO $_2$  is significantly improved by adding just 1% Y $_2$ O $_3$ . Besides, NO $_2$  adsorption and desorption time ( $\tau_{ads}$  and  $\tau_{des}$ ) decreased to 75% and 50% of that of undoped sample, respectively.

**Key words:** WO<sub>3</sub>, Y<sub>2</sub>O<sub>3</sub>, NO<sub>2</sub>, Sensitivity, response time.

Nitrogen Oxide NO<sub>x</sub> (NO<sub>2</sub> and NO) is toxic to human body and is an air pollutant and a source for acidic rain so that high sensitivity sensors with fast response time are required. Tungsten Oxide is an n-type semiconductor and is a good candidate for NO<sub>2</sub> detection [1] and have been successfully used as low-cost gas sensors for air quality measurement [2]. Doping of pure metal oxides with metals or oxides has shown promising results to overcome their inherent limitations on sensor performance [3].

To fabricate doped low-cost Tungsten Oxide sensors, WO $_3$  bulk powder was ball milled with different amounts of  $Y_2O_3$  (1, 5 and 10 wt%) in Isopropanol in a planetary ball milling machine for 2 hours at 350 rpm. 10 µm thick film gas sensitive resistors were fabricated by screen-printing of WO $_3$  and  $Y_2O_3$  surface decorated WO $_3$  fine grains on a 2x2x0.250 mm alumina sensor substrate with an inter-digitated gold electrode on front side and a platinum heater circuit on the back side and were oven annealed at 800  $^{\circ}$ C in air. Electrical gas responses of fabricated sensors exposed to 200 ppb NO $_2$  were measured at temperature range 250-450  $^{\circ}$ C.

XRD patterns presented in Fig. 1 show the evolution of  $Y_2O_3$  peaks in doped samples. The morphologies of bulk WO<sub>3</sub> and  $Y_2O_3$ -WO<sub>3</sub> doped

samples are illustrated in Fig. 2. It is shown that the surface of WO<sub>3</sub> grains are well decorated with Y<sub>2</sub>O<sub>3</sub> nanoparticles and the decoration percentage improves by increasing the amount of Y<sub>2</sub>O<sub>3</sub> with an optimum of 5%. Also, it seems that in sample with 10% Y<sub>2</sub>O<sub>3</sub>, WO<sub>3</sub> grain growth are a bit restricted. Electrical gas sensing measurements of Y<sub>2</sub>O<sub>3</sub> doped WO<sub>3</sub> to 200 ppb NO<sub>2</sub> are presented in Fig.3. As shown in fig.3a, the sensitivity of each film (relative response, R<sub>q</sub>/R<sub>a</sub>) decreases by increasing the operating temperature, besides, it is obvious that by adding only 1 wt% Y<sub>2</sub>O<sub>3</sub> to WO<sub>3</sub> the RR significantly increase, and this improvement continuous by increasing the amount of Y2O3 to 5 wt% and stops, so that the RR in Y<sub>2</sub>O<sub>3</sub> 5% and 10% are almost equal, confirming the doping yield of 5%, also see Table 1. Among different operating temperatures, samples showed faster response and recovery times at 350°C. Fig. 3b represents the normalized electrical responses of doped samples to 200 ppb NO2 at 350°C. As can be seen, adding 1% Y<sub>2</sub>O<sub>3</sub> has greatly enhanced the RR response and decreased adsorption and desorption times ( $\tau_{ads}$  &  $\tau_{des}$ ) to 75% and 50% of that of undoped WO<sub>3</sub>, see table 1. Increasing the amount of dopant gradually increases the RR but depresses the response time.

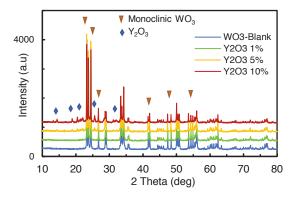


Fig. 1. XRD patterns of WO<sub>3</sub>, WO<sub>3</sub>-Y<sub>2</sub>O<sub>3</sub> 1%, WO<sub>3</sub>-Y<sub>2</sub>O<sub>3</sub> 5% and WO<sub>3</sub>-Y<sub>2</sub>O<sub>3</sub> 10%

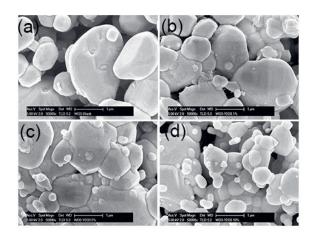
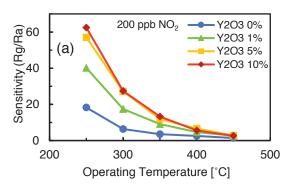


Fig. 2. SEM image of printed films; a) undoped WO<sub>3</sub>, b) WO3-1% Y<sub>2</sub>O<sub>3</sub>, c) WO3-5% Y<sub>2</sub>O<sub>3</sub>, d) WO3-10% Y<sub>2</sub>O<sub>3</sub>



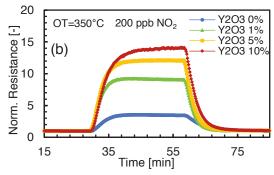


Fig. 3. a) Sensitivity of blank and doped WO<sub>3</sub> screen-printed sensors to 200 ppb NO<sub>2</sub> at temperature range 250-450°C, b) Electrical response to 200 ppb NO<sub>2</sub> at 350°C

Table 1- Sensing characteristics of Y<sub>2</sub>O<sub>3</sub> doped WO<sub>3</sub> films in terms of Relative Response (RR), and response times (τ<sub>ads</sub> and τ<sub>des</sub>) to 200 ppb NO<sub>2</sub> at 350 °C.

OT (°C)	Sample	WO₃ (wt%)	Y <sub>2</sub> O <sub>3</sub> (wt%)	RR=Rg/Ra [-]	τ <sub>ads</sub> (min)	τ <sub>des</sub> (min)
	Y <sub>2</sub> O <sub>3</sub> 0%	100	0	3.566	8.01	9.65
350	Y <sub>2</sub> O <sub>3</sub> 1%	99	1	9.091	6.08	4.73
	Y <sub>2</sub> O <sub>3</sub> 5%	95	5	11.937	7.12	6.25
	Y <sub>2</sub> O <sub>3</sub> 10%	90	10	13.288	8.33	7.12

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