# Synthesis and Enhanced Gas Sensing Properties of MoS<sub>2</sub> Nanoflakes

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#### **Abstract**

Morphology and structure control is an efficient way to improve the sensing performance of sensing materials. In this work,  $MoS_2$  nanoflakes were synthesized via a facile hydrothermal route without any template. The sensing measurements reveal that the  $MoS_2$  nanoflakes not only exhibit high response to  $NO_2$  but also reduce the working consumption. The improved sensing properties are mainly attributed to the few layer nanoflakes structure and high specific surface area, which provides more active edge and sufficient room for reactions.

# Introduction

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MoS<sub>2</sub>, a p-type semiconductor, received increasing attention because its promising layered structure which facilitates intercalation of molecules and ions between layers, and exhibits potential application in gas sensor devices[1]. Although some team began to study the sensing property of MoS<sub>2</sub> in recent years, the high operating temperature and relative low sensitivity are still the major limitation achieving widespread for applications[2]. One method to improve the properties of gas sensors is controlling the dimension and morphology of nanomaterials to enlarge the surface to volume ratio[3]. In this work, we report a facile way to produce MoS<sub>2</sub> nanoflakes which show high performance for NO<sub>2</sub> detection at room temperature.

MoS<sub>2</sub>,

nanoflakes,

# **Experimental**

MoS $_2$  nanoflakes were synthesized via a simple hydrothermal method by using Na $_2$ MoO $_4$ ·2H $_2$ O and L-cysteine as reactant, the pH of the solution was modulated by oxalate acid. The hydrothermal process maintained for 48 h under 200 $^{\circ}$ C. The different molar ratio of Mo and S make the different morphology of production, and the Mo/S was controlled by the molar ratio of Na $_2$ MoO $_4$ ·2H $_2$ O and L-cysteine The samples were named 1-MoS $_2$  and 3-MoS $_2$  for Mo/S=1 and 3, respectively.

The samples were pasted on interdigitated Au electrodes previously patterned on SiO<sub>2</sub> for

further gas sensing test. The sensing properties were detected by a homemade sensing test system.

temperature

# Results and discussion

sensing,

# Microstructure

gas

Figure 1 shows that the morphologies and structures of  $1\text{-MoS}_2$  and  $3\text{-MoS}_2$  samples. The morphology of these two  $\text{MoS}_2$  samples changed from nanospheres to nanoflakes as the Mo/S ratio changed from 1 to 3. The lattice fringes in two samples with a constant spacing of 0.610 nm ascribed to the (002) plane of  $\text{MoS}_2$ , which illustrate that the Mo/S ratio only change the morphology of  $\text{MoS}_2$ .

# Gas sensing property

Figure 2a present the sensing response of these two samples to 300 ppm  $NO_2$  at different temperatures. The result indicated that the 3-MoS<sub>2</sub> exhibited the excellent sensing property (189.52%) at room temperature, while the sensitivity of 1-MoS<sub>2</sub> reached maximum(25.8%) at 150 °C . The dynamic response of 3-MoS<sub>2</sub> under different concentration from 10-400 ppm of  $NO_2$  at room temperature was revealed in Figure 2b, which illustrate that this sensor exhibited excellent response and recovery characteristic to different concentration of  $NO_2$ .

 ${\mbox{MoS}_2}$  nanoflakes shows its p-type semiconductor character in this case,  ${\mbox{NO}_2}$  molecules capture electrons from  ${\mbox{MoS}_2}$  conductance band, which thicken the hole accumulation layer and decrease the resistance

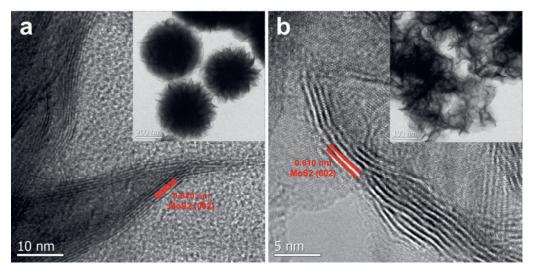


Figure 1: The morphology and structure of (a)1-MoS<sub>2</sub> nanospheres and (b) 3-MoS<sub>2</sub>

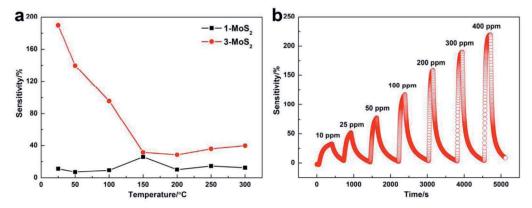


Figure 2: (a) The responses of  $1-MoS_2$  and  $3-MoS_2$  working at different temperature to 300 ppm  $NO_2$ . (b) Dynamic response of  $3-MoS_2$  nanosheets at room temperature.

of the sensors[4]. A proposed chemical interaction mechanism is given by Eq.(1).

$$MoS2+NO2(gas) = MoS2+NO2(ads)+h+ (1)$$

The enhancement in gas sensing properties on 3-MoS<sub>2</sub> nanoflakes were attributed to the high surface area. Comparing with the 1-MoS<sub>2</sub> nanoflakes, 3-MoS<sub>2</sub> nanoflakes were looser, the sheets are thinner which exposed increased active edge of MoS<sub>2</sub> and provide more room for chemical reaction with gas.

#### **Conclusions**

In summary, changing the molar ratio of  $Na_2MoO_4$  and L-cysteine is a efficient way to control the morphology of  $MoS_2$  nanostructures, from nanospheres to nanoflakes. The sensor based on the 3- $MoS_2$  nanoflakes increasing the sensitivity(189.52%) as well as reducing the working temperature comparing with 1- $MoS_2$  nanospheres. The enhancement of sensing properties are ascribed to the thinner flakes and large specific surface area which increased active edge and reaction rooms of 3- $MoS_2$  nanoflakes.

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