

Pulsed polarization on Au|YSZ NO_x-sensors with and without catalytic layer

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Summary:

Sensors of the type Au|YSZ|Au were produced, and some of them were covered with an additional Pt containing catalyst layer. These sensors were operated in the pulsed polarization mode. Their sensitivity to NO_x was investigated. The pure gold sensor showed almost no NO signal, but a clear signal towards NO₂. The catalyst-coated sensor responded to both gases, NO and NO₂. Therefore, we assume that NO₂ is essential for the sensor effect, which could not have been explicitly shown in previous studies.

Keywords: pulsed polarization, Au | YSZ, NO_x detection, exhaust gas sensor, dynamic method

Motivation

Gaseous components from combustion processes are among the main pollutants in our environment. Therefore, it is important to measure them for continuous optimization of the combustion process. Newer approaches for the detection of such gases are directed towards dynamic instead of static methods. This is expected to lead to further improvements in selectivity. Examples for such methods are cyclovoltammetry [1], thermocyclic operation [2] or pulsed polarization [3]. In the latter case, nitrogen oxides could be detected selectively, but the exact mechanism is still unclear. In this method, catalytically active Pt-electrodes are polarized at 400 °C and the subsequent self-discharge behavior is evaluated. It is expected that NO and NO₂ are present in thermodynamic equilibrium at the electrode, independent of the actual concentrations [4]. In this work, the influences of the NO/NO₂-equilibrium on the sensor signal will be investigated in more detail.

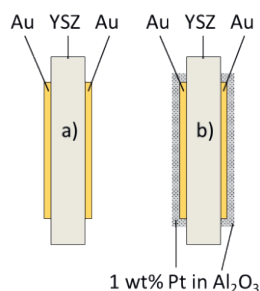


Fig. 1. Schematic view of the sensors a) without and b) with catalytic layer

Experimental

The here-used sensors consist of screen-printed Au-electrodes (area = 5 x 5 mm²) that are applied to both sides of an 8YSZ substrate and fired at 850 °C (see Fig. 1). A catalytic layer was then added to some of the electrodes. It consists of 1 wt% Pt added to porous Al₂O₃. These catalytic layers were fired at 700 °C, whereby the sensors without catalytic layer were also fired in order to avoid influences by the firing process.

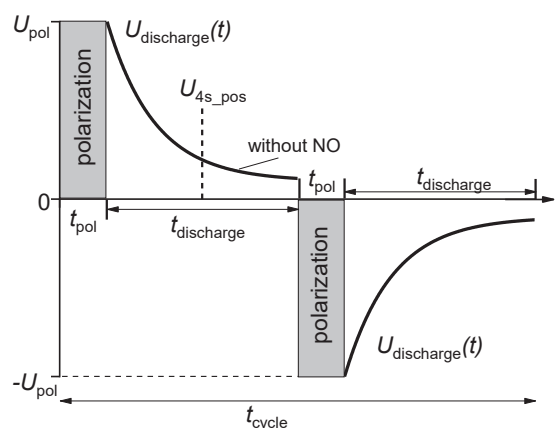


Fig. 2. Overview of a pulsed polarization cycle

Then, the sensors were operated by pulsed polarization. For this purpose, the sensor was polarized for a polarization duration $t_{pol} = 1$ s with a polarization voltage $U_{pol} = 1$ V. Afterwards the self-discharge was measured as an open circuit potential, OCP, for $t_{discharge} = 10$ s. These polarization and self-discharge phases were repeated continuously, always polarizing with alternating

polarity. To generate a sensor signal from these cycles, the discharge voltage is evaluated at a fixed point in time in the cycle, for example 4 s after positive polarization U_{4s_pos} (Fig. 2).

A mixture containing 10% O_2 and 2% H_2O in N_2 was used as base gas. Additionally, 50–400 ppm NO and NO_2 were added stepwise. All measurements were performed at 400 °C in a tube furnace.

Results

The results of two measurements of Au|YSZ|Au are shown in Fig. 3. Fig 3a shows the discharge voltages of an Au-sensor without catalyst layer 1 s and 9 s after positive polarization. It can be seen that this sensor almost does not respond to NO. Thus, on the 1 s curve almost no NO influence is visible. After 9 s a slight change to more positive voltages can be seen, which means a slowing down of the self-discharge. With added NO_2 , however, clear change to more negative voltages can be seen. This indicates an accelerated self-discharge. These NO_2 signals are visible after 1 s as well as after 9 s and show a similar effect as observed for sensors with Pt-electrodes [3].

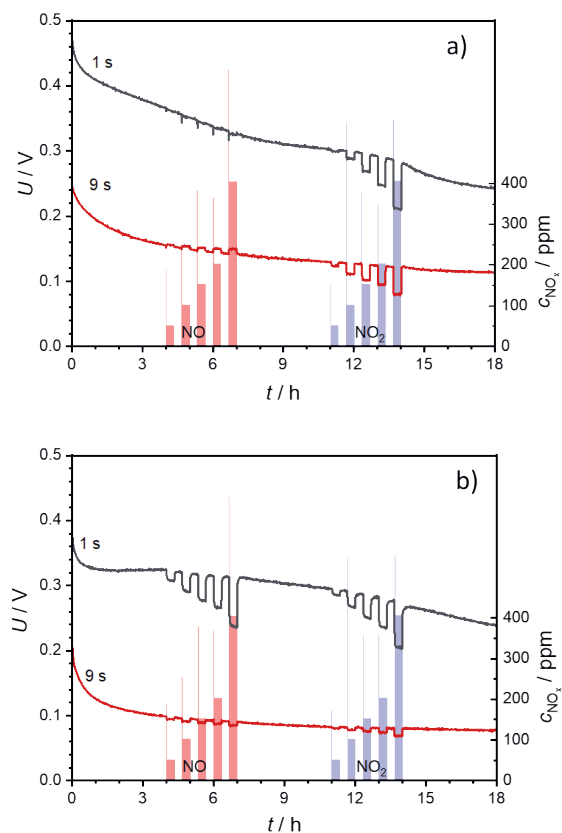


Fig. 3. Depolarization voltages after 1 and 9 s after positive polarization at Au|YSZ|Au sensors a) without and b) with additional catalytic layer on both electrodes.

Fig. 3b shows the discharge curves of a sensor with a catalytic layer on top of both gold electrodes. Here, a clear signal for NO as well as for NO_2 can be seen. This is particularly pronounced 1 s after polarization. Here, the effects of NO and NO_2 hardly differ.

Discussion

These results show that an Au sensor without catalyst hardly responds to NO but clearly responds to NO_2 . Only with a catalytic layer, which oxidizes a part of the NO to NO_2 due to the thermodynamic equilibrium, NO becomes measurable. Since the electrodes of Pt sensors are catalytically active themselves, they may also react primarily to NO_2 , which is formed by oxidation of NO directly at the electrode.

Conclusion

It could be shown that NO_2 in contrast to NO at 400 °C has a direct influence on the self-discharge of the sensor. In order to further investigate this effect, measurements with single-sided catalytic coated electrodes as well as measurements at other temperatures will be performed.

References

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