

Planar Bragg Grating Sensors Functionalized with Cyclodextrins for Trichlorofluoromethane Sensing

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Summary:

This contribution summarizes on recent findings of planar Bragg grating sensors, functionalized with cyclodextrins, for the detection of ozone depleting trichlorofluoromethane. Detection limits as low as 5 ppm are feasible whereas sensitivity and dynamic depend on the employed cyclodextrin class. The prospect of transferring the technology from silicas to polymer-based devices is also presented.

Keywords: Optical Sensor, Bragg Grating, Evanescent Field, Cyclodextrin, Trichlorofluoromethane

Introduction

Once an auspicious chlorofluorocarbon, widely employed as propellant and refrigerant, the usage and fabrication of trichlorofluoromethane, also referred to as CFC-11 or R-11, is now strictly prohibited due to its vast ozone depletion potential [1]. Nevertheless, the substance is still released nowadays, for example during the recycling of obsolete cooling devices or even in illegal production plants [2].

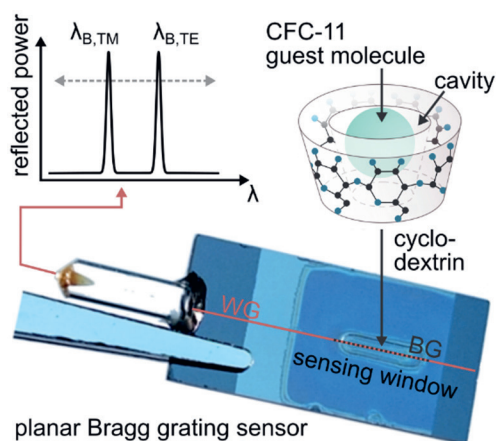


Fig. 1. Working principle of a planar Bragg grating sensor functionalized for CFC-11 sensing via cyclodextrin coatings.

Thus, reliable and sensitive detection of CFC-11 is still of vital importance for the society and yet continuously a technological challenge for modern sensors. Optical Bragg gratings constitute a promising technology for this task, since, besides low weight, they offer outstanding electromagnetic, chemical and thermal resistance.

However, they necessitate functionalization for the detection of CFC-11, which can be achieved by coating the sensitive Bragg grating region with cyclodextrins (CDs). Due to their molecular structure and composition, CDs are able to form a non-covalent host-guest complex with CFC-11, as illustrated in Fig. 1. Coating a planar Bragg grating (PBG) device with CDs enables quantification of the CFC-11 molecule abundance via the evanescent field interaction of guided mode and functional coating, which leads to a shift of the PBG's modal Bragg reflection peaks $\lambda_{B,TE}$ and $\lambda_{B,TM}$. Based on their composition, CDs are classified as α -, β - or γ -cyclodextrin. Further modification of its solubility, viscosity and selectivity, is adapted by substitution of the CD's hydroxyl groups.

Sensor Response

An overview of the employed CD derivatives is given in Tab. 1, while Fig. 2 depicts the respective Bragg wavelength shift $\Delta\lambda_B$ of both modal reflection peaks as a function of the CFC-11 content, diluted in nitrogen. It is found that, in all cases, the maximum response of the TE reflection peak is about ten times larger than that of the TM signal. *Per*-methyl substituted derivatives exhibit maximum wavelength shifts up to

Tab. 1: *Per*-substituted cyclodextrin derivatives.

Substitute	α -CD	β -CD	γ -CD
Methyl	CD1	CD2	CD3
Ethyl	CD4	CD5	CD6
Allyl	CD7	CD8	CD9

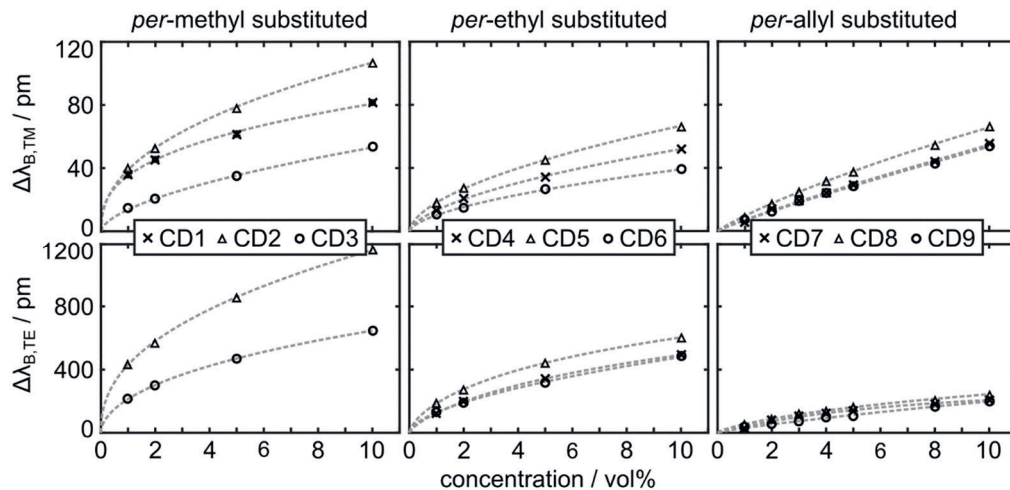


Fig. 2. Bragg wavelength shift of both modal (TE & TM) reflection peaks as a function of CFC-11 concentration in nitrogen, for α -, β - or γ -cyclodextrin derivative coatings with various hydroxy substitutions.

1200 pm (CD2), which results in a detection limit of 5 ppm. This value is about 400 times larger than that of an uncoated PBG. The signal deflection of *per*-ethyl- and *per*-allyl substituted derivatives, however, is reduced. Albeit, in contrast to the determined sensitivities, these derivatives show a significantly faster temporal response when the sensor is exposed to CFC-11 in nitrogen. For example, the rise times for CD2, CD5 and CD8, at 1 vol% CFC-11, are 1435 s, 71 s and 45 s, respectively. Consequently, it is feasible to tailor the functionalized PBG's behavior by employing the appropriate CD coating [3].

Spontaneous Crystallization

Exposing a PBG coated with CD1 to a CFC-11 content of at least 35 vol% leads to spontaneous crystallization of CFC-11 on the surface of the PPBG, as depicted in Fig. 3.

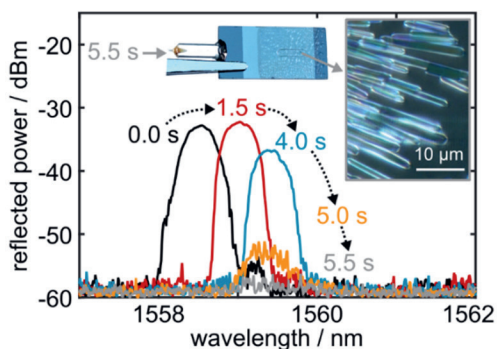


Fig. 3. Signal loss due to spontaneous crystallization. Inset: Sensor with crystallized surface and microscopic image thereof.

Within a timeframe of 5.5 s, this leads to complete signal loss due to a drastic refractive index increase of the functionalization coating and / or scattering losses by the structural reconfiguration of the crystallized surface [4].

Conclusion and Outlook

In conclusion, PBGs functionalized with CDs are well-suited for the detection of volatile trichlorofluoromethane, whereas sensor sensitivity and response time can be adapted by employing appropriate CD derivatives. CFC-11 quantities above 35 vol% lead to spontaneous crystallization of the coating which can be exploited for the development of new filter and storage concepts [5]. While all PBGs used in this study are SiO₂ based, future research will focus on transferring the demonstrated methodology on polymer-based planar devices and the development of new affinity materials.

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