

Carbon-Encapsulated Co₃O₄ Nanofibers for Room Temperature Selective NH₃ sensors

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Abstract :

Carbon-encapsulated Co₃O₄ nanofiber composites with rGO were successfully fabricated by using the electrospinning technology. The morphology and resistance of Carbon-encapsulated Co₃O₄ samples can be controlled by preparing condition. The as-electrospun carbon-Co₃O₄ nanofiber composites based chemo resistive sensor all showed good sensitivity to different concentrations of ammonia. Besides this, the sensor exhibited good selectivity to several potential interferents such as methanol, ethanol, formaldehyde, acetone, benzene, and methylbenzene. Water did play an important role in the sensing process. We discussed the sensing mechanism based on DFT-based first principle calculations.

Key words: Electrospinning, Carbon encapsulated Co₃O₄ Nanofibers, Room-Temperature Ammonia Sensors, DFT Calculations

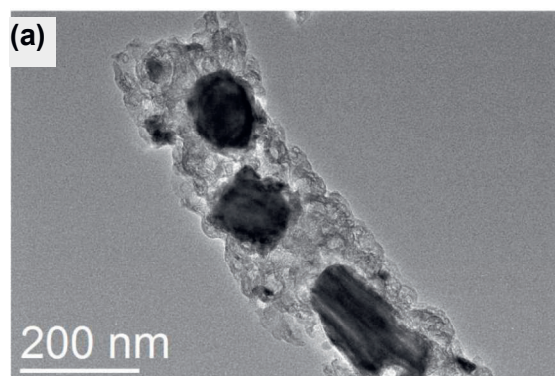
Introduction

The rGO with noble metals/metal oxide/organic ligands nanocomposites have been found to effectively improve ammonia gas sensing properties because of their potentials to combine the desirable properties of different nanoscale building blocks to improve mechanical, chemical and electronic properties [1]. As mean while, the first principles were more and more widely used in studying the atomic and electronic structure informations of materials[2]. I. Maity et al. studied the selectivity tuning of graphene oxide based gas sensor devices, where the role of oxygen functional groups were investigated for hysisorption of NO₂, NH₃, CO, and H₂O using first principle calculation. Besides, reduced graphene oxide (rGO) encapsulated Co₃O₄ composite nanofibers had shown excellent sensing properties in our previous work[3]. It is necessary to further study the preparing condition that affected the morphology and resistance, and discuss the sensing mechanism based on experimental results and DFT-based first principle calculations.

Results and conclusions

The morphology and resistance of carbon-encapsulated Co₃O₄ samples were affected by the existence of a small amount of O₂ during the thermal treatment. The morphology of

carbon-encapsulated Co₃O₄ samples of A0 and A1 was shown as the SEM images in Fig. 1. A small amount of O₂ was added in the thermal treatment for sample A1 which had resistance of several hundreds of thousand Ohm. For A0 which had resistance of only several Ohm, we did not add O₂ for purpose. The sensors of A1 showed good sensitivity to ammonia (see Fig. 2.) while the sensors of A0 did not. Such desirable response features may be attributable to the unique interaction of ammonia with the porous structure with the polarized C-Co³⁺ covalent centers.



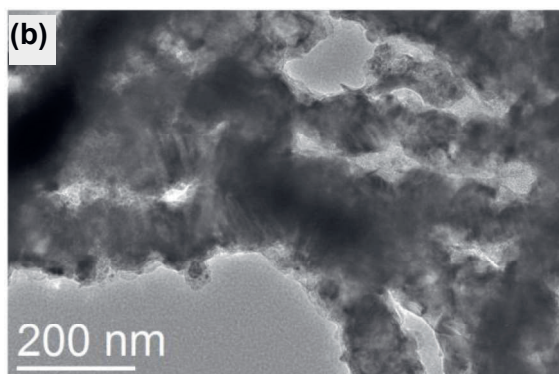


Fig. 1. Images of carbon-encapsulated Co_3O_4 samples: (a) no O_2 was added for purpose (b) a small amount of O_2 was added during the thermal treatment.

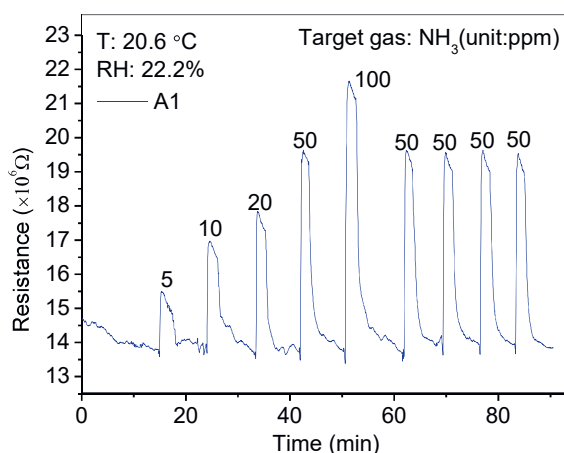


Fig. 2. The response to NH_3 of 5-100 ppm and the repeatability of response to NH_3 of 50 ppm for A1.

Water did play an important role in the sensing process which was shown in the results of sensing test (see Fig. 3).

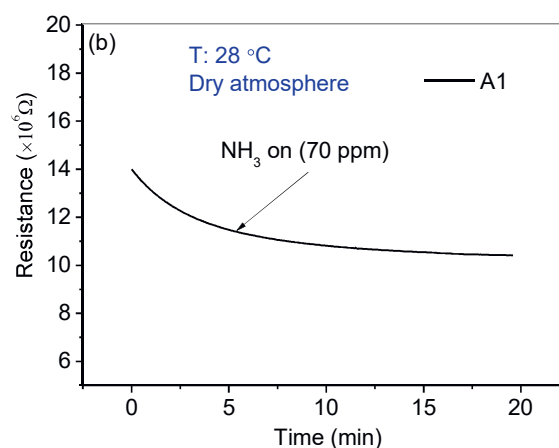
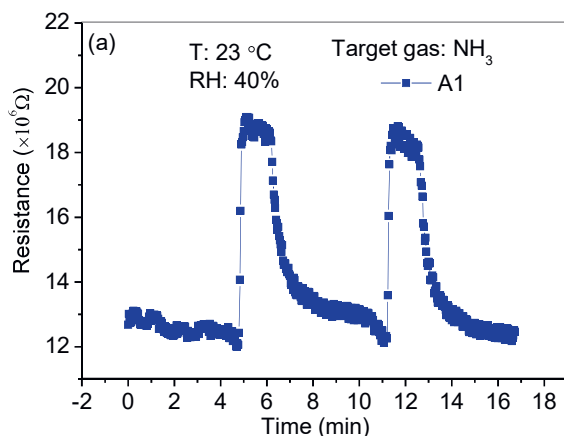


Fig. 3. The response to NH_3 of 70 ppm for (a) in moist air with 40% RH and (b) in dry atmosphere.

Among various volatile organic compounds, the carbon-encapsulated Co_3O_4 nanofibers showed excellent selectivity for ammonia to several possible interferences as shown in Fig. 4.

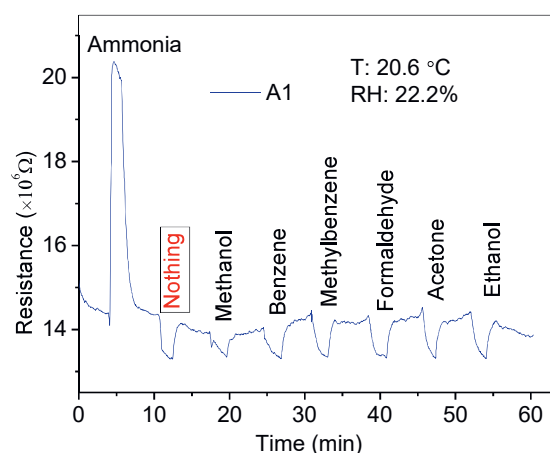


Fig. 4. Selectivity of the response of carbon-encapsulated Co_3O_4 composite nanofibers.

Finally, we also discussed the sensing mechanism based on DFT-based first principle calculations.

References

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